

Palladium-Catalyzed S_N2'-Cyclization of Ambivalent (Bromoalkadienyl)malonates: Preparation of Medium- to Large-**Membered Endocyclic Allenes**

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Supporting Information

ABSTRACT: A palladium-catalyzed reaction for preparing various endocyclic allenes was developed. The substrates for the reaction were readily available ω -(pronucleophile-tethered)-3-bromo-1,3-alkadienes, and a palladium-catalyst facilitated their unimolecular S_N2'-cyclization in the presence of potassium tert-butoxide to give the corresponding 9- to 16membered endocyclic allenes in fair yields of up to 67% together with the dimeric 16- to 32-membered endocyclic bis-

HNu
$$\rightarrow$$
 Br $(CH_2)_n$ Nu \rightarrow $(CH_2)_n$ Nu \rightarrow

allenes and other oligomeric/polymeric intermolecular reaction products. For higher yields of the monomeric endocyclic allenes, the reaction needed to be conducted under high-dilution conditions. Using a chiral palladium catalyst, axially chiral endocyclic allenes were obtained in up to 70% ee.

■ INTRODUCTION

Allenes are a class of compounds characterized by two cumulated carbon-carbon double bonds. Because of their propadiene structures, they show distinctive steric and electronic properties and have emerged as highly interesting target molecules in organic synthesis. 1,2 Incorporation of an allenic substructure into a carbocycle creates a peculiar topological property.3 Although various endocyclic allenes have been prepared and reported so far, most of their synthetic methods construct an allenic motif within a preformed carbocycle precursor.4-6 Two of the most widely applicable methods in the synthesis of endocyclic allenes, namely the ring enlargement of cycloalkenes via the corresponding gemdibromobicyclo[n.1.0]alkenes by the Doering-Moore-Skattebøl reaction⁴ and dehydrohalogenation of 1-halocycloalkenes,⁵ are also classified in this category. Another approach to the endocyclic allenes is cyclization of preformed acyclic allenes. This approach, however, is not applicable to the synthesis of relatively smaller ring cyclic allenes. The third category of the endocyclic allene synthesis is simultaneous construction of both allenic and cyclic structures by a single process, but the reported examples in this category are very rare.8 Since 2000, we have developed a palladium-catalyzed reaction of preparing functionalized allenes starting with 2-halo 1,3-dienes and appropriate soft nucleophiles (eq 1 in Scheme 1). The reaction was extended into an asymmetric counterpart by the use of a chiral Pd catalyst, and enantiomerically enriched axially chiral allenes were obtained in up to 94% ee. 10 We also demonstrated that the palladium-catalyzed reaction could be

Scheme 1. Palladium-Catalyzed Reaction of 2-Halo 1,3-Dienes with a Soft Nucleophile

used for the synthesis of a series of endocyclic allenes starting with cycloalkenones.5

In this report, we describe a novel method of preparing various endocyclic allenes utilizing the Pd-catalyzed reaction. The two reactants of the palladium-catalyzed reaction, namely a 2-bromo-1,3-diene and a soft nucleophile, are incorporated into a single molecule, and the S_N2'-cyclization of the "ambivalent" substrates proceeds primarily intramolecularly to afford various endocyclic allenes in fair yields.

The allenic substructure, which is with two cumulated orthogonal carbon-carbon double bonds, suppresses the relative motion among the five consecutive atoms (marked in red; eq 2 in Scheme 1). Similarly, the relative orientation of the five atoms is fastened in the (alkylidene- π -allyl)palladium

Received: May 17, 2017 Published: June 22, 2017 intermediate of the palladium-catalyzed reaction. ¹¹ These rigid substructures in the palladium intermediates as well as in the allenic products may lead to the unusual preference/selectivity in the present cyclization reactions. Indeed, the cyclization process could provide various medium to large carbocycles in reasonable yields. It should be mentioned that formation of five- to seven-membered carbocycles, which are easy to assemble in typical cyclization processes, are impractical by the present method because thermodynamically stable endocarbocyclic allenes are usually nine-membered or larger. ¹² Although a few kinetically stabilized eight-membered carbocyclic allenes were reported to be isolable, ¹³ endocyclic allenes with a seven-membered or a smaller carbocycle exist only as transient reactive intermediates.

■ RESULTS AND DISCUSSION

Preparation of Pronucleophile-Tethered 3-Bromo 1,3-Diene Derivatives 3 and 4. The substrates for our endocyclic allene synthesis are pronucleophile-tethered 3-bromo 1,3-dienes (3a-h and 4f), and they are prepared from ω -bromoalkanals by the simple three-step sequence as depicted in Scheme 2. Readily available ω -bromoalkanals were converted

Scheme 2. Preparation of Pronucleophile-Tethered 3-Bromo 1,3-Dienes

$$\begin{array}{c} \text{Br} & \text{CHO} \\ \hline \text{CH}_2\text{Cl}_2, \ 0 \ ^\circ\text{C} \\ \hline & \textbf{1a-h} \\ \hline & \textbf{1a-h} \\ \hline & \textbf{(61-77\%)} \\ \end{array} \\ \textbf{a-g}, \ n = 3-9; \ \textbf{h}, \ n = 11 \\ \textbf{3}, \ \text{Nu} = C(\text{CO}_2\text{Me})_2 \\ \hline & \textbf{0}_2\text{S} \\ \end{array} \\ \textbf{4}, \ \text{Nu} = \begin{array}{c} C \\ \hline \text{CH}_2\text{Cl}_2, \ 0 \ ^\circ\text{C} \\ \hline \textbf{1a-h} \\ \hline & \textbf{Br} \\ \hline & \textbf{1a-h} \\ \hline & \textbf{1a-h} \\ \hline & \textbf{Br} \\ \hline & \textbf{1a-h} \\$$

into the corresponding 1,1, ω -tribromoalkenes 1a-i in 61-77% yields by the standard Wittig dibromoolefination (Ramirez olefination). The palladium-catalyzed cross-coupling of 1 with vinylzinc chloride proceeded at the sterically less congested alkenyl-Br (i.e., C=CBr trans to the ω -bromoalkyl group) selectively to give the corresponding (Z)-3, ω -dibromo-1,3-dienes 2a-h in 50-68% yields. Alkylation of dimethyl malonate or benzodithiole tetraoxide (BDT) with 2 furnished designed substrates 3a-h and 4f in pure form (70-95%).

Palladium-Catalyzed Reactions of Pronucleophile-Tethered 3-Bromo 1,3-Diene Derivatives 3 and 4. Substrates 3 and 4 prepared as in Scheme 2 were subjected to the palladium-catalyzed reaction. At the outset, the conditions for the palladium-catalyzed reaction were optimized using 3e (n = 7) as a model substrate, and the results of the optimization studies are summarized in Table 1. The reaction of 3e was efficiently catalyzed by 5 mol % of a palladium complex generated in situ from Pd₂(dba)₄ and dpbp¹⁶ (1.1 equiv to Pd) in THF in the presence of potassium tert-butoxide (1.1 equiv to 3e). Under the rather standard conditions, i.e., with the initial concentration of substrate 3e being 1.4×10^{-2} mol/L, 3e was completely consumed within 20 h at 30 °C. However, to our disappointment, the yield of the desired 12membered endocyclic allene 5e was only 9% (entry 1). Complete consumption of 3e and the low yield of 5e indicated that intermolecular processes giving oligomeric/polymeric products competed with the unimolecular cyclization reaction. Indeed, reactions under dilute conditions facilitated the desired intramolecular pathway leading to the higher yields of 5e (entries 2 and 3). When the initial concentration of 3e was 1.4 \times 10⁻³ mol/L, **5e** was isolated in 31% yield (entry 3). It should be mentioned that the reactions in entries 1-3 produced dimeric endocyclic bis-allene 6e in 11-19% yields together with 5e. Endocyclic bis-allene 6e was obtained as a mixture of

Table 1. Optimization of Reaction Conditions for Palladium-Catalyzed S_N2'-Cyclization of 3e^a

entry	solvent	base	ligand	concn of 3e (mol/L)	yield of 5e (%)	yield of $6e^b$ (%)
1	THF	KO^tBu	dpbp	1.4×10^{-2}	9 ^c	13 ^c
2	THF	KO^tBu	dpbp	1.4×10^{-2}	22 ^c	11 ^c
3	THF	KO^t Bu	dpbp	4.6×10^{-3}	31 ^c	19^c
4	DMF	KO^tBu	dpbp	1.4×10^{-3}	8^d	nd^e
5	toluene	KO^tBu	dpbp	1.4×10^{-3}	7^d	nd^e
6	CH_2Cl_2	KO^tBu	dpbp	1.4×10^{-3}	0^d	nd^e
7	THF	NaOMe	dpbp	1.4×10^{-3}	31 ^c	15 ^c
8	THF	CsO^tBu	dpbp	1.4×10^{-3}	16 ^d	nd^e
9	THF	Cs_2CO_3	dpbp	1.4×10^{-3}	0	nd^e
10	THF	NaOMe	DPEphos	1.4×10^{-3}	23^d	nd^e
11	THF	NaOMe	dppf	1.4×10^{-3}	25^d	nd^e

^aThe reaction was carried out with 3f (140 μmol) and a base (154 μmol) in the given solvent at 30 °C for 20 h in the presence of a palladium catalyst (5 mol %) generated from $Pd_2(dba)_4$ and a bisphosphine ligand. ^bObtained as dl/meso-diastereomeric mixtures. ¹⁷ Fisolated yield by recycle HPLC. ^aDetermined by GC analysis. ^eNot determined.

The Journal of Organic Chemistry

Table 2. Influences of Ring Size in Palladium-Catalyzed S_N2' Cyclization of 3^a

yield of 6^{b-d} (%) yield of 7^{b-d} (%) substrate 3 yield of 5^b (%) entry 1 3a (n = 3)0 (5a) 30 (6a) 10 (7a) 2 3b (n = 4)23 (5b) 16 (6b) 3 3c (n = 5)11 (5c) 24 (6c) 7 (7c) 3d (n = 6)18 (5d) 27 (6d) 9 (7d) 4 5 3e(n = 7)31 (5e) 19 (6e) 6 3f(n = 8)52 (5f) 14 (6f) 3g (n = 9)42 (5g) 21 (6g) 3h (n = 11)33 (5h) 10 (6g)

^aThe reaction was carried out with 3 (350 μ mol) and a base (385 μ mol) in THF (250 mL) at 30 °C for 20 h in the presence of a palladium catalyst (5 mol %) generated from Pd₂(dba)₄ and dpbp. ^bIsolated yield by recycle HPLC. ^cCalculated on the basis of 3. ^dObtained as diastereomeric mixtures. ¹⁷ ^eNot determined or <5% yield.

dl- and *meso*-diastereomers with respect to the allenic axial chirality; two allenic central sp-C signals were detected at δ 205.8 and 205.9 in the 13 C NMR spectrum (see the Experimental Section). THF was the best solvent examined so far. Reactions in DMF, toluene, or dichloromethane gave **5e** in much lower yields ranging from 0 to 8% (entries 4–6). The choice of a proper base drastically influences the yield of **5e**; while the reactions with KO'Bu or NaOMe afforded the monomeric endocyclic allene in 31% yield (entries 3 and 7), the yield of **5e** declined to 16% with the use of CsO'Bu (entry 8), and the reaction using Cs₂CO₃ did not provide **5e** at all (entry 9). On the other hand, the effects of the bisphosphine ancillary ligands were minor, and the reactions with DPEphos 18 (1.1 equiv to Pd) or dppf gave **5e** in 23% and 25% yields, respectively (entries 10 and 11).

The reaction conditions shown in entry 3 of Table1 were selected as "optimized conditions", and they were applied to the other substrates in the same way. The results of the palladium-catalyzed reaction are summarized in Table 2.

While the substrates with more than three methylene units (i.e., 3b and the longer ones) afforded the corresponding monomeric endocyclic allenes 5b-h in moderate yields by the palladium-catalyzed reaction, the reaction of 3a did not provide eight-membered endocyclic allene 5a (entry 1). The highest yield of 5 was achieved for the reaction of 3f giving 13membered monomeric endocyclic allene 5f in 52% yield (entry 6). With a longer polymethylene chain in 3g (n = 9) and 3h (n = 9) = 11), the yields of 5g (42%) and 5h (33%) were slightly lower (entries 7 and 8). Likewise, shorter substrates 3b-e afforded the corresponding monomeric endocyclic allenes 5b-e in lower yields ranging from 11% to 31% (entries 2-5). In all cases including the reaction of 3a, substrates 3 were completely consumed under these conditions, and the formation of dimeric endocyclic bis-allenes 6, which were isolated in 10-30% yields by the HPLC separation, was observed. Homologous trimeric endocyclic tris-allenes 7a, 7c, and 7d were also isolated in 10%, 7%, and 9% yields, respectively, in the reactions giving 6 in

relatively higher yields (entries 1, 3, and 4). Since bis-allenes 6 and tris-allenes 7 possess multiple allenic axially chiral elements, they were obtained as diastereomeric mixtures.¹⁷

Whereas eight-membered endocarbocyclic allenes are generally too reactive to be isolated without appropriate kinetically stabilizing bulky substituents, 3a,e,13 it was rational that 5a was not detected in the reaction of 3a. However, the absence of 5a should not be ascribed to its high reactivity. Once highly reactive 5a was generated in situ, it was expected to undergo [2+2] cycloaddition 19 giving an isomeric mixture of homodimers such as 8, but 8 and/or related isomeric species were not detected in the reaction mixture. The key intermediate of the palladium-catalyzed reaction of 3a is (alkylidene- π -allyl) palladium species 9. Due to the short trimethylene chain in 9, the nucleophilic malonate moiety is not capable of attacking the CH_2 π -allyl terminus intramolecularly, and thus, the formation of 5a could not be realized (Scheme 3).

The cyclization reaction of 4f was also examined. Substrate 4f possesses a benzo[1,3]dithiole tetraoxide (BDT) substructure as a pronucleophilic moiety. The palladium-catalyzed reaction of 4f was conducted essentially in the same way as in Table 2

Scheme 3

$$\begin{array}{c|c} CO_2Me & Pd^0, base \\ \hline MeO_2C & Pd^0, base \\ \hline \end{array}$$

The Journal of Organic Chemistry

Scheme 4. Palladium-Catalyzed S_N2' Cyclization of BDT-Tethered Substrate 4f

Table 3. Palladium-Catalyzed Asymmetric Synthesis of Axially Chiral Endocyclic Allenes 5f/10f^{ct}

on here	substrate	chiral ligand	base	solvent	yield ^b (%)	% ee (config) ^{c,d}
entry	substrate	chirai figand	Dase	solvent	yield (%)	
1	3f	(R)-L1	KO ^t Bu	THF	50 (5f)	65 (S)
2	3f	(R)-L1	KO^tBu	dioxane	5 (5f)	11 (S)
3	3f	(R)-L1	KO^t Bu	CH_2Cl_2	nd^e	
4	3f	(R)-L1	NaOMe	THF	13 (5f)	59 (S)
5	3f	(R)-L1	CsO^tBu	THF	35 (5f)	55 (S)
6	3f	(R)-L2	KO^tBu	THF	42 (5f)	45 (S)
7	3f	(R)-L3	KO^tBu	THF	25 (5f)	70 (S)
8	3f	(R)- L4	KO^t Bu	THF	34 (5f)	55 (S)
9	3f	(R)-L5	KO^tBu	THF	54 (5f)	50 (S)
10	4f	(R)-L3	KO ^t Bu	THF	29 (10f)	51 (S)

"The reaction was carried out with 3f or 4f (100 μ mol) and a base (110 μ mol) in a given solvent (70 mL) at 30 °C for 60 h in the presence of a palladium catalyst (5 mol %) generated from Pd₂(dba)₄ and a chiral ligand L. ^bIsolated yield by recycle HPLC. ^cDetermined by chiral HPLC analysis (see the Experimental Section for details). ^dThe absolute configurations were deduced by the Lowe–Brewster rule. ²⁰ ^eNot determined.

except for the reaction time. Due probably to the lower nucleophilicity of BDT, the reaction took longer time (120 h) for completion, and monomeric endocyclic allene 10f was obtained in the highest yield of 67% (Scheme 4). The formation of the corresponding endocyclic dimeric bis-allene and trimeric tris-allene was minor (<5%) in this reaction. The BDT moiety in 4f is sterically more compact than the malonate moiety in 3f, which may facilitate the intramolecular cyclization in the palladium-catalyzed process.

Palladium-Catalyzed Asymmetric Synthesis of Axially Chiral Endocyclic Allenes 5f and 10f. Endocyclic allenes 5 and 10 obtained as in Table 2 and Scheme 4 are axially chiral but racemic. Using 3f as a representative substrate, palladiumcatalyzed enantioselective synthesis of 5f was examined according to our previous studies; 10 that is, with a palladium catalyst (5 mol %) generated from Pd₂(dba)₄ and (R)-segphos ((R)-L1) in THF at 30 °C in the presence of KO^tBu, (+)-5f was obtained in 50% yield with 65% ee (Table 3, entry 1). Among the various solvents and bases examined, the combination of THF and KOtBu showed the highest yield and the highest enantioselectivity for the reaction of 3f (entries 1-5). While (R)-L2, (R)-L4, and (R)-L5 showed lower enantioselectivity than (R)-L1 with 45-55% ee (entries 6, 8, and 9), the enantioselectivity was improved to 70% ee by the use of (R)-L3 although the chemical yield of (+)-5f was

considerably lower (entry 7). The enantioselective reaction of 4f was conducted as in entry 7, and (+)-10f was obtained in 51% ee and 29% yield (entry 10). The absolute configurations of dextrorotatory (+)-5f and (+)-10f were deduced to be (*S*) by the Lowe–Brewster rule.²⁰

CONCLUSIONS

In summary, we have developed a general method of preparing various endocyclic allenes of a 9- to 16-membered carbocycle utilizing the palladium-catalyzed reaction. Readily available ω -(pronucleophile-tethered)-3-bromo-1,3-alkadienes 3 and 4 undergo the unimolecular S_N2'-cyclization in the presence of an appropriate palladium catalyst to give the corresponding endocyclic allenes 5/10 in fair yields together with the dimeric endocyclic bis-allenes and higher oligomers/polymers. The highest yield of 67% was achieved for the preparation of the 13membered cyclic allene. Due probably to the rigid substructures in the allenic C=C=C products as well as in the (alkylidene- π -allyl)palladium intermediates, the medium to large-membered carbocyclic compounds are accessible in reasonable yields by the present cyclization reaction. By the use of a chiral palladium catalyst, axially chiral endocyclic allenes were obtained in up to 70% ee.

■ EXPERIMENTAL SECTION

General Methods. All anaerobic and/or moisture-sensitive manipulations were carried out with standard Schlenk techniques under predried nitrogen or with glovebox techniques under prepurified argon. ¹H NMR (at 400 MHz) and ¹³C NMR (at 100 MHz) chemical shifts are reported in ppm downfield of internal tetramethylsilane. The HRMS measurements were carried out by the EI method with a TOF analyzer. Tetrahydrofuran (from benzophenone-ketyl) and dichloromethane (from CaH₂) were distilled under nitrogen prior to use. Pd(PPh₃)₄, ¹²Pd₂(dba)₄, ²²dpbp, ¹⁶(R)-L1, ²³(R)-L2, ²⁴(R)-L4, ²⁵(R)-L5, ²⁶ and 1,3-benzodithiole-1,1,3,3-tetraoxide (BDT)²⁷ were prepared according to the reported methods. All of the other chemicals were obtained from commercial sources and used as received unless otherwise noted.

Preparation of ω-Bromoalkanals. The ω-bromoalkanals were prepared from the corresponding commercially available ω-bromo-1-alkanols according to the reported procedure, ²⁸ in which the preparation of 8-bromooctanal (n=7 in Scheme 2) was described in detail. All the ω-bromoalkanals used in this study are known compounds and were characterized by comparison of their NMR spectra with those reported previously (n=3, ²⁹ n=4, ²⁹ n=5, ³⁰ n=6, ³¹ n=7, ²⁸ n=8, ³² n=9, ³² n=11, ³³). **Preparation of 1,1,ω-Tribromo-1-alkenes (1a–h).** To a

Preparation of 1,1,ω-Tribromo-1-alkenes (1a–h). To a CH₂Cl₂ (ca. 2 mL/ ω -bromoalkanal 1 mmol) solution of CBr₄ (1.5 equiv to ω -bromoalkanal) was added PPh₃ (3.0 equiv to ω -bromoalkanal) portionwise at 0 °C, and the solution was allowed to stir at this temperature for 20 min. A solution of ω -bromoalkanal (1 equiv) in CH₂Cl₂ (ca. 1 mL/ ω -bromoalkanal 2 mmol) was added dropwise at 0 °C, and the mixture was stirred at the same temperature for 10 min. After the reaction mixture was quenched with a small amount of water (ca. 1 mL), the mixture was concentrated under reduced pressure. Addition of hexane (ca. 5 mL/ ω -bromoalkanal 1 mmol) to the concentrated solution precipitated triphenylphosphine oxide as pale yellow solid, which was removed by filtration. The filtrate was evaporated, and the residue was purified by silica gel column chromatography (with hexane). The characterization data of 1a—h are given below.

1,1,5-Tribromo-1-pentene (1a; n=3). Colorless liquid. Yield: 6.27 g (63%) starting with 4-bromobutanal (4.90 g; 32.5 mmol). 1 H NMR (CDCl₃): δ 1.96–2.03 (m, 2H), 2.25–2.31 (m, 2H), 3.42 (t, J=6.6 Hz, 2H), 6.40 (t, J=7.3 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 30.8, 31.7, 32.5, 90.5, 136.7. EI-HRMS: calcd for C₅H₇Br₃ 303.8098, found 303.8101

1,1,6-Tribromo-1-hexene (1b; n = 4). Colorless liquid. Yield: 5.78 g (61%) starting with 5-bromopentanal (4.90 g; 29.7 mmol). 1 H NMR (CDCl₃): δ 1.56–1.64 (m, 2H), 1.86–1.93 (m, 2H), 2.12–2.17 (m, 2H), 3.42 (t, J = 6.7 Hz, 2H), 6.39 (t, J = 7.2 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 26.3, 31.9, 32.1, 33.3, 89.5, 137.9. EI-HRMS: calcd for C₆H₉Br₃ 317.8254, found 317.8251.

1,1,7-Tribromo-1-heptene (1c; n = 5). Colorless liquid. Yield: 3.96 g (65%) starting with 6-bromohexanal (3.26 g; 18.2 mmol). ¹H NMR (CDCl₃): δ 1.41–1.52 (m, 4H), 1.84–1.91 (m, 2H), 2.09–2.14 (m, 2H), 3.41 (t, J = 6.8 Hz, 2H), 6.39 (t, J = 7.3 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.0, 27.6, 32.5, 32.8, 33.6, 89.1, 138.3. EI-HRMS: calcd for $C_7H_{11}Br_3$ 331.8411, found 331.8405.

1,1,8-Tribromo-1-octene (1d; n = 6). Colorless liquid. Yield: 1.78 g (61%) starting with 7-bromoheptanal (1.62 g; 8.39 mmol). ¹H NMR (CDCl₃): δ 1.31–1.38 (m, 2H), 1.41–1.49 (m, 4H), 1.83–1.90 (m, 2H), 2.07–2.13 (m, 2H), 3.41 (t, J = 6.8 Hz, 2H), 6.38 (t, J = 7.3 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.6, 27.9, 28.2, 32.7, 32.9, 33.9, 88.8, 138.6. EI-HRMS: calcd for C₈H₁₃Br₃ 345.8567, found 345.8570.

1,1,9-Tribromo-1-nonene (1e; n=7). Colorless liquid. Yield: 5.64 g (77%) starting with 8-bromooctanal (4.18 g; 20.2 mmol). ¹H NMR (CDCl₃): δ 1.31–1.47 (m, 8H), 1.82–1.89 (m, 2H), 2.07–2.12 (m, 2H), 3.41 (t, J=6.8 Hz, 2H), 6.38 (t, J=7.2 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.7, 28.1, 28.5, 28.8, 32.7, 32.9, 34.0, 88.7, 138.7. EI-HRMS: calcd for $C_0H_{15}Br_3$ 359.8724, found 359.8718.

1,1,10-Tribromo-1-decene (1f; n = 8). Colorless liquid. Yield: 4.66 g (63%) starting with 9-bromononanal (4.34 g; 19.6 mmol). ¹H NMR

(CDCl₃): δ 1.28–1.47 (m, 10H), 1.82–1.89 (m, 2H), 2.06–2.12 (m, 2H), 3.41 (t, J = 6.8 Hz, 2H), 6.39 (t, J = 7.3 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 27.8, 28.1, 28.7, 28.9, 29.2, 32.8, 33.0, 34.1, 88.6, 138.9. EIHRMS: calcd for C_{10} H $_{17}$ Br $_{3}$ 373.8880, found 373.8881.

1,1,11-Tribromo-1-undecene (1g; n=9). Colorless liquid. Yield: 3.06 g (66%) starting with 10-bromodecanal (2.79 g; 11.9 mmol). 1 H NMR (CDCl₃): δ 1.26–1.43 (m, 12H), 1.82–1.89 (m, 2H), 2.06–2.12 (m, 2H), 3.41 (t, J=6.9 Hz, 2H), 6.38 (t, J=7.2 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 27.8, 28.1, 28.7, 29.0, 29.2, 29.3, 32.8, 33.0, 34.1, 88.5, 138.9. EI-HRMS: calcd for C₁₁H₁₉Br₃ 387.9037, found 387.9037.

1,1,13-Tribromo-1-tridecene (1h; n = 11). Colorless liquid. Yield: 4.68 g (68%) starting with 12-bromododecanal (4.32 g; 16.4 mmol).

¹H NMR (CDCl₃): δ 1.26–1.46 (m, 16H), 1.82–1.89 (m, 2H), 2.06–2.11 (m, 2H), 3.41 (t, J = 6.9 Hz, 2H), 6.38 (t, J = 7.2 Hz, 1H).

¹³C{¹H} NMR (CDCl₃): δ 27.8, 28.2, 28.8, 29.1, 29.3, 29.4, 29.49, 29.50, 32.9, 33.0, 34.1, 88.5, 138.9. EI-HRMS: calcd for C₁₃H₂₃Br₃ 415.9350, found 415.9356.

Preparation of (Z)-3,\omega-Dibromo-1,3-alkadienes (2a–h). To a suspension of (CH₂=CH)ZnCl in THF, prepared from dry ZnCl₂ (28 mmol) and vinylmagnesium chloride (2.0 M THF solution; 12 mL; 24 mmol), was added a solution of 1,1, ω -tribromo-1-alkene 1 (8.0 mmol) and Pd(PPh₃)₄ (0.20 mmol) in THF (16 mL) at 0 °C. After the mixture was stirred for 4.5 h at room temperature, it was diluted with hexane, filtered, and evaporated to dryness. The residue was purified by silica gel chromatography (with hexane) and further purified by recycle HPLC to give the title compound. The characterization data of **2a–h** are given below.

(*Z*)-3,7-Dibromo-1,3-heptadiene (**2a**; n=3). Colorless liquid. Yield: 1.10 g (54%). ¹H NMR (CDCl₃): δ 1.99–2.06 (m, 2H), 2.46–2.52 (m, 2H), 3.43 (t, J=6.8 Hz, 2H), 5.20 (d, J=10.4 Hz, 1H), 5.56 (d, J=16.3 Hz, 1H), 5.98 (t, J=7.2 Hz, 1H), 6.32 (dd, J=16.3 and 10.4 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 30.3, 31.5, 32.9, 118.2, 127.4, 132.7, 135.7. EI-HRMS: calcd for $C_7H_{10}Br_2$ 251.9149, found 251.9144.

(*Z*)-3,8-Dibromo-1,3-octadiene (**2b**; n=4). Colorless liquid. Yield: 1.16 g (54%). 1 H NMR (CDCl $_3$): δ 1.58–1.66 (m, 2H), 1.87–1.94 (m, 2H), 2.34–2.39 (m, 2H), 3.43 (t, J=6.7 Hz, 2H), 5.19 (d, J=10.5 Hz, 1H), 5.55 (d, J=16.3 Hz, 1H), 5.97 (t, J=7.1 Hz, 1H), 6.31 (dd, J=16.3 and 10.5 Hz, 1H). 13 C{ 1 H} NMR (CDCl $_3$): δ 26.9, 30.6, 32.2, 33.6, 117.7, 126.6, 134.0, 135.7. EI-HRMS: calcd for C $_8$ H $_{12}$ Br $_2$ 265.9306, found 265.9303.

(*Z*)-3,9-Dibromo-1,3-nonadiene (**2c**; n=5). Colorless liquid. Yield: 1.17 g (52%). ¹H NMR (CDCl₃): δ 1.47–1.50 (m, 4H), 1.85–1.92 (m, 2H), 2.31–2.36 (m, 2H), 3.41 (t, J=6.8 Hz, 2H), 5.17 (d, J=10.5 Hz, 1H), 5.54 (d, J=16.3 Hz, 1H), 5.98 (t, J=7.2 Hz, 1H), 6.31 (dd, J=16.3 and 10.5 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.5, 27.8, 31.3, 32.6, 33.7, 117.5, 126.2, 134.5, 135.8. EI-HRMS: calcd for C₉H₁₄Br₂ 279.9462, found 279.9467.

(*Z*)-3,10-Dibromo-1,3-decadiene (**2d**; n=6). Colorless liquid. Yield: 1.21 g (51%). 1 H NMR (CDCl₃): δ 1.32–1.51 (m, 6H), 1.83–1.90 (m, 2H), 2.30–2.35 (m, 2H), 3.41 (t, J=6.9 Hz, 2H), 5.17 (d, J=10.5 Hz, 1H), 5.53 (d, J=16.3 Hz, 1H), 5.97 (t, J=7.1 Hz, 1H), 6.31 (dd, J=16.3 and 10.5 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 28.0, 28.2, 28.4, 31.4, 32.7, 34.0, 117.3, 126.1, 134.9, 135.8. EI-HRMS: calcd for C_{10} H $_{16}$ Br $_2$ 293.9619, found 293.9617.

(*Z*)-3,11-Dibromo-1,3-undecadiene (**2e**; n=7). Colorless liquid. Yield: 1.56 g (63%). ¹H NMR (CDCl₃): δ 1.31–1.49 (m, 8H), 1.82–1.89 (m, 2H), 2.29–2.34 (m, 2H), 3.41 (t, J = 6.9 Hz, 2H), 5.16 (d, J = 10.5 Hz, 1H), 5.53 (d, J = 16.3 Hz, 1H), 5.98 (t, J = 7.1 Hz, 1H), 6.31 (dd, J = 16.3 and 10.5 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 28.1, 28.2, 28.6, 29.0, 31.5, 32.8, 34.0, 117.2, 126.0, 135.0, 135.8. EI-HRMS: calcd for C₁₁H₁₈Br₂ 307.9775, found 307.9777.

(*Z*)-3,12-Dibromo-1,3-dodecadiene (**2f**; n=8). Colorless liquid. Yield: 1.30g (50%). 1 H NMR (CDCl₃): δ 1.26–1.46 (m, 10H), 1.82–1.89 (m, 2H), 2.28–2.34 (m, 2H), 3.41 (t, J=6.8 Hz, 2H), 5.16 (d, J=10.5 Hz, 1H), 5.53 (d, J=16.3 Hz, 1H), 5.98 (t, J=7.1 Hz, 1H), 6.31 (dd, J=16.3 and 10.5 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 28.2, 28.3, 28.7, 29.15, 29.24, 31.5, 32.8, 34.1, 117.2, 125.9, 135.2, 135.9. EI-HRMS: calcd for $C_{12}H_{20}Br_2$ 321.9932, found 321.9931.

(*Z*)-3,13-Dibromo-1,3-tridecadiene (**2g**; n=9). Colorless liquid. Yield: 1.84 g (68%). 1 H NMR (CDCl₃): δ 1.30–1.34 (m, 8H), 1.39–1.46 (m, 4H), 1.82–1.89 (m, 2H), 2.28–2.34 (m, 2H), 3.41 (t, J=6.9 Hz, 2H), 5.16 (d, J=10.5 Hz, 1H), 5.53 (d, J=16.3 Hz, 1H), 5.98 (t, J=7.1 Hz, 1H), 6.31 (dd, J=16.3 and 10.5 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 28.1, 28.3, 28.7, 29.2, 29.28, 29.32, 31.5, 32.8, 34.1, 117.1, 125.8, 135.2, 135.9. EI-HRMS: calcd for C_{13} H₂₂Br₂ 336.0088, found 336.0091.

(*Z*)-3,15-Dibromo-1,3-pentadecadiene (**2h**; n=11). Colorless liquid. Yield: 1.64 g (56%). 1 H NMR (CDCl₃): δ 1.28–1.47 (m, 16H), 1.82–1.89 (m, 2H), 2.28–2.34 (m, 2H), 3.41 (t, J=6.9 Hz, 2H), 5.15 (d, J=10.5 Hz, 1H), 5.52 (d, J=16.3 Hz, 1H), 5.98 (t, J=7.1 Hz, 1H), 6.31 (dd, J=16.3 and 10.5 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 28.2, 28.4, 28.8, 29.3, 29.450, 29.455, 29.53 (2C), 31.6, 32.9, 34.1, 117.1, 125.8, 135.3, 135.9. EI-HRMS: calcd for $C_{15}H_{26}Br_2$ 364.0401, found 364.0404.

Preparation of Malonate-Tethered Bromodienes 3a–h. A general procedure is given below. To a solution of NaH (4.9 mmol) in DMF (10 mL) was added dimethyl malonate (4.9 mmol) at 0 °C. After gas evolution ceased, dibromoalkadiene 2 (4.0 mmol) was added to the solution by means of syringe under nitrogen. The mixture was stirred at room temperature for 20 h. After being quenched with saturated NH₄Cl_{aq}, the reaction mixture was extracted with Et₂O twice and the combined organic layer was washed with H₂O and NaCl_{aq}, dried over MgSO₄, filtered, and evaporated under reduced pressure. The residue was purified by silica gel chromatography (hexane/ethyl acetate = 4/1) to afford the malonate-tethered bromodiene 3. The characterization data of 3a–h are given below.

Dimethyl 2-((Z)-5-Bromo-4,0-heptadienyl)-1,3-propanedioate (**3a**; n=3). Colorless liquid. Yield: 854 mg (70%). ¹H NMR (CDCl₃): δ 1.43–1.53 (m, 2H), 1.92–1.98 (m, 2H), 2.33–2.38 (m, 2H), 3.39 (t, J=7.5 Hz, 1H), 3.74 (s, 6H), 5.18 (d, J=10.4 Hz, 1H), 5.54 (d, J=16.3 Hz, 1H), 5.96 (t, J=7.1 Hz, 1H), 6.30 (dd, J=16.3 and 10.4 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 26.1, 28.4, 31.1, 51.5, 52.6, 117.7, 126.6, 133.8, 135.7, 169.8. ESI-HRMS: calcd for C₁₂H₁₇BrO₄Na 327.0208 (M + Na), found 327.0220. Anal. Calcd for C₁₂H₁₇BrO₄: C, 47.23; H, 5.61. Found: C, 47.23; H, 5.57.

Dimethyl 2-((Z)-6-Bromo-5,7-octadienyl)-1,3-propanedioate (**3b**; n=4). Colorless liquid. Yield: 1.02 g (80%). ¹H NMR (CDCl₃): δ 1.32–1.40 (m, 2H), 1.45–1.52 (m, 2H), 1.90–1.96 (m, 2H), 2.30–2.35 (m, 2H), 3.37 (t, J=7.5 Hz, 1H), 3.74 (s, 6H), 5.17 (d, J=10.5 Hz, 1H), 5.53 (d, J=16.3 Hz, 1H), 5.96 (t, J=7.3 Hz, 1H), 6.31 (dd, J=16.3 and 10.5 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.0, 27.9, 28.7, 31.2, 51.6, 52.6, 117.5, 126.3, 134.5, 135.8, 169.9. EI-HRMS: calcd for C₁₃H₁₉BrO₄: 318.0467, found 318.0468. Anal. Calcd for C₁₃H₁₉BrO₄: C, 48.92; H, 6.00. Found: C, 48.77; H, 5.97.

Dimethyl 2-((Z)-7-Bromo-6,8-nonadienyl)-1,3-propanedioate (**3c**; n = 5). Colorless liquid. Yield: 1.00 g (75%). ¹H NMR (CDCl₃): δ 1.29–1.50 (m, 6H), 1.88–1.93 (m, 2H), 2.28–2.33 (m, 2H), 3.36 (t, J = 7.6 Hz, 1H), 3.74 (s, 6H), 5.16 (d, J = 10.6 Hz, 1H), 5.52 (d, J = 16.3 Hz, 1H), 5.96 (t, J = 7.1 Hz, 1H), 6.31 (dd, J = 16.3 and 10.6 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.1, 27.9, 28.69, 28.71, 31.3, 51.6, 52.4, 117.2, 126.0, 134.7, 135.8, 169.9. Anal. Calcd for C₁₄H₂₁BrO₄: C, 50.46; H, 6.35. Found: C, 50.37; H, 6.29. ESI-HRMS: calcd for C₁₄H₂₁BrO₄Na 355.0515 (M + Na), found 355.0516.

Dimethyl 2-((Z)-8-Bromo-7,9-decadienyl)-1,3-propanedioate (**3d**; n=6). Colorless liquid. Yield: 1.01 g (73%). ¹H NMR (CDCl₃): δ 1.29–1.34 (m, 6H), 1.40–1.47 (m, 2H), 1.87–1.93 (m, 2H), 2.28–2.33 (m, 2H), 3.36 (t, J=7.5 Hz, 1H), 3.74 (s, 6H), 5.16 (d, J=10.3 Hz, 1H), 5.53 (d, J=16.3 Hz, 1H), 5.97 (t, J=7.2 Hz, 1H), 6.31 (dd, J=16.3 and 10.3 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.3, 28.2, 28.8, 28.9, 29.0, 31.5, 51.7, 52.5, 117.2, 126.0, 135.0, 135.9, 170.0. EI-HRMS: calcd for C₁₅H₂₃BrO₄ 346.0780, found 346.0781. Anal. Calcd for C₁₅H₂₃BrO₄: C, 51.88; H, 6.68. Found: C, 51.72; H, 6.60.

Dimethyl 2-((Z)-9-Bromo-8,10-undecadienyl)-1,3-propanedioate (**3e**; n = 7). Colorless liquid. Yield: 1.37 g (95%). ¹H NMR (CDCl₃): δ 1.26–1.32 (m, 8H), 1.39–1.47 (m, 2H), 1.87–1.92 (m, 2H), 2.28–2.33 (m, 2H), 3.36 (t, J = 7.6 Hz, 1H), 3.74 (s, 6H), 5.16 (d, J = 10.4 Hz, 1H), 5.53 (d, J = 16.3 Hz, 1H), 5.97 (t, J = 7.1 Hz, 1H), 6.31 (dd, J = 16.3 and 10.4 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.3, 28.3, 28.8,

29.09, 29.11 (2C), 31.5, 51.7, 52.5, 117.2, 125.9, 135.2, 135.9, 170.0. EI-HRMS: calcd for $C_{16}H_{25}BrO_4$ 360.0936, found 360.0936. Anal. Calcd for $C_{16}H_{25}BrO_4$: C, 53.19; H, 6.97. Found: C, 52.66; H, 6.91.

Dimethyl 2-((Z)-10-Bromo-9,11-dodecadienyl)-1,3-propane-dioate (3f; n=8). Colorless liquid. Yield: 1.20 g (80%). ¹H NMR (CDCl₃): δ 1.26–1.33 (m, 10H), 1.39–1.47 (m, 2H), 1.87–1.92 (m, 2H), 2.28–2.33 (m, 2H), 3.36 (t, J=7.6 Hz, 1H), 3.74 (s, 6H), 5.16 (d, J=10.5 Hz, 1H), 5.53 (d, J=16.4 Hz, 1H), 5.98 (t, J=7.1 Hz, 1H), 6.31 (dd, J=16.4 and 10.5 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.4, 28.3, 28.9, 29.18, 29.21, 29.22, 29.3, 31.5, 51.8, 52.5, 117.2, 125.9, 135.2, 135.9, 170.0. EI-HRMS: calcd for C₁₇H₂₇BrO₄ 374.1093, found 374.1093. Anal. Calcd for C₁₇H₂₇BrO₄: C, 54.41; H, 7.25. Found: C, 54.48; H, 7.27.

Dimethyl 2-((Z)-11-Bromo-10,12-tridecadienyl)-1,3-propane-dioate (3g; n=9). Colorless liquid. Yield: 1.42 g (91%). ¹H NMR (CDCl₃): δ 1.28–1.33 (m, 12H), 1.40–1.47 (m, 2H), 1.87–1.92 (m, 2H), 2.28–2.33 (m, 2H), 3.36 (t, J=7.6 Hz, 1H), 3.74 (s, 6H), 5.15 (d, J=10.4 Hz, 1H), 5.52 (d, J=16.3 Hz, 1H), 5.98 (t, J=7.3 Hz, 1H), 6.31 (dd, J=16.3 and 10.4 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.3, 28.3, 28.8, 29.16, 29.21, 29.25, 29.34, 29.38, 31.5, 51.7, 52.4, 117.1, 125.8, 135.2, 135.9, 170.0. EI-HRMS: calcd for C₁₈H₂₉BrO₄: C, 55.53; H, 7.51. Found: C, 55.60; H, 7.39.

Dimethyl 2-((Z)-13-Bromo-12,14-pentadecadienyl)-1,3-propanedioate (3h; n=11). Colorless liquid. Yield: 1.29 g (77%). ¹H NMR (CDCl₃): δ 1.26–1.35 (m, 16H), 1.40–1.47 (m, 2H), 1.87–1.92 (m, 2H), 2.28–2.34 (m, 2H), 3.36 (t, J=7.6 Hz, 1H), 3.74 (s, 6H), 5.15 (d, J=10.4 Hz, 1H), 5.52 (d, J=16.3 Hz, 1H), 5.98 (t, J=7.1 Hz, 1H), 6.31 (dd, J=16.3 and 10.4 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 27.4, 28.4, 28.9, 29.2, 29.29, 29.32, 29.4, 29.51, 29.54, 29.6, 31.6, 51.8, 52.5, 117.1, 125.8, 135.3, 135.9, 170.0. ESI-HRMS: calcd for C₂₀H₃₃BrO₄Na 439.1454 (M + Na), found 439.1449. Anal. Calcd for C₂₀H₃₃BrO₄: C, 57.55; H, 7.97. Found: C, 57.69; H, 7.84.

2-((Z)-10-Bromo-9,11-dodecadienyl)-1,3-benzodithiole 1,1,3,3-Tetraoxide (4f). To a mixture of NaH (26 mg, 1.1 mmol) and BDT (240 mg, 1.1 mmol) in DMF (4.5 mL) was added dibromododecadiene 2f (274 mg, 844 μ mol) by means of syringe under nitrogen. After the mixture was stirred for 16 h at 80 °C, it was extracted with ethyl acetate twice, and the combined organic layer was washed with H₂O and NaCl_{aq}, dried over MgSO₄, filtered, and evaporated under reduced pressure. The residue was purified by recycle HPLC to afford 4f (263 mg, 570 mol, 68%) as a white solidMp: 67-69 °C. ¹H NMR $(CDCl_3)$: δ 1.33–1.50 (m, 10H), 1.75–1.80 (m, 2H), 2.30–2.34 (m, 4H), 4.36 (t, J = 7.1 Hz, 1H), 5.16 (d, J = 10.4 Hz, 1H), 5.53 (d, J = 10.4 Hz, 1H), 5.53 (d, J = 10.4 Hz, 1H), 5.53 16.3 Hz, 1H), 5.99 (t, J = 7.0 Hz, 1H), 6.32 (dd, J = 16.3 and 10.4 Hz, 1H) 7.91-7.94 (m, 2H), 8.02-8.05 (m, 2H). ¹³C{¹H} NMR (CDCl₃): δ 22.1, 25.8, 28.2, 28.8, 29.00, 29.04, 29.1, 31.4, 73.7, 117.1, 122.5, 125.8, 135.1, 135.78, 135.79, 137.6. EI-HRMS: calcd for C₁₉H₂₅BrO₄S₂ 460.0378, found 460.0367. Anal. Calcd for C₁₉H₂₅BrO₄S₂: C, 49.46; H, 5.46. Found: C, 49.80; H, 5.40.

Palladium-Catalyzed Reaction of Malonate-Tethered Bromodienes 3a—h. A general procedure is given below. To a mixture of $Pd_2(dba)_4$ (17.5 μ mol), dpbp (19.2 μ mol), and KO^tBu (385 μ mol) in THF (250 mL) was added malonate-tethered bromodiene 3 (350 μ mol, 1.4 mM) by means of syringe under nitrogen. After the mixture was stirred for 20 h at 30 °C, it was concentrated, filtered through a short pad of silica gel, and evaporated to dryness. The crude product was purified by recycle HPLC to afford the corresponding endocyclic allenes 5, 6, and 7. The characterization data of the cyclic allenic products (5, 6, and 7) are listed below.

Dimethyl 2,2-(2,3-Octadiene-1,8-diyl)-1,3-propanedioate (**5b**; n = 4). Colorless liquid. Yield: 19.2 mg (23%). ¹H NMR (CDCl₃): δ 1.42–1.58 (m, 4H), 1.93–2.06 (m, 3H), 2.36–2.46 (m, 1H), 2.57–2.63 (m, 1H), 2.81–2.87 (m, 1H), 3.71 (s, 3H), 3.72 (s, 3H), 4.94–5.01 (m, 2H). ¹³C{¹H} NMR (CDCl₃): δ 18.9, 24.1, 24.8, 28.4, 31.6, 52.5, 52.6, 57.0, 85.3, 87.4, 172.2, 172.6, 209.8. ESI-HRMS: calcd for C₁₃H₁₈O₄Na: 261.1103 (M + Na), found 261.1106.

Dimethyl 2,2-(2,3-Nonadiene-1,9-diyl)-1,3-propanedioate (**5c**; n = 5). Colorless liquid. Yield: 9.7 mg (11%). 1 H NMR (CDCl₃): δ 1.29–1.84 (m, 7H), 1.88–1.96 (m, 1H), 2.04–2.11 (m, 1H), 2.32–

2.41 (m, 1H), 2.61–2.72 (m, 2H), 3.72 (s, 3H), 3.73 (s, 3H), 4.98–5.05 (m, 1H), 5.29–5.32 (m, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 20.0, 20.6, 28.3, 29.7, 29.8, 32.6, 52.5, 52.6, 55.5, 87.0, 92.9, 172.0, 172.7, 205.9. EI-HRMS: calcd for C₁₄H₂₀O₄ 252.1362, found 252.1363.

Dimethyl 2,2-(2,3-Decadiene-1,10-diyl)-1,3-propanedioate (**5d**; n = 6). Colorless liquid. Yield: 16.8 mg (18%). ¹H NMR (CDCl₃): δ 1.18–1.62 (m, 8H), 1.78–1.96 (m, 2H), 2.04–2.15 (m, 2H), 2.59 (dd, J = 14.1 and 11.6 Hz, 1H), 2.79 (dt, J = 14.1 and 4.4 Hz, 1H), 3.71 (s, 3H), 3.74 (s, 3H), 5.01–5.08 (m, 1H), 5.26–5.31 (m, 1H). ¹³C{¹H} NMR (CDCl₃): δ 20.4, 26.0, 26.1, 26.4, 26.8, 30.9, 34.0, 52.5, 52.6, 57.2, 87.4, 91.4, 171.8, 172.2, 206.0. EI-HRMS: calcd for $C_{15}H_{22}O_4$ 266.1518, found 266.1516.

Dimethyl 2,2-(2,3-Undecadiene-1,11-diyl)-1,3-propanedioate (**5e**; n=7). Colorless liquid. Yield: 30.4 mg (31%). ¹H NMR (CDCl₃): δ 0.97–1.59 (m, 10H), 1.74–1.82 (m, 1H), 1.93–2.17 (m, 3H), 2.55 (dd, J=13.9 and 11.6 Hz, 1H), 2.68–2.73 (m, 1H), 3.71 (s, 3H), 3.74 (s, 3H), 4.61–4.67 (m, 1H), 4.89–4.96 (m, 1H). ¹³C{¹H} NMR (CDCl₃): δ 19.1, 21.4, 22.3, 25.9, 26.48, 26.54, 27.3, 32.1, 52.5, 52.6, 56.8, 83.8, 89.2, 171.7, 171.8, 208.4. EI-HRMS: calcd for C₁₆H₂₄O₄ 280.1675, found 280.1675. Anal. Calcd for C₁₆H₂₄O₄: C, 68.54; H, 8.63. Found: C, 68.28; H, 8.74.

Dimethyl 2,2-(2,3-Dodecadiene-1,12-diyl)-1,3-propanedioate (5f; n=8). Colorless liquid. Yield: 53.6 mg (52%). ¹H NMR (CDCl₃): δ 1.06–1.15 (m, 1H), 1.26–1.62 (m, 11H), 1.90–2.12 (m, 4H), 2.55 (dd, J=14.6 and 11.4 Hz, 1H), 2.77 (dtd, J=14.6, 4.4, and 1.2 Hz, 1H), 3.71 (s, 3H), 3.73 (s, 3H), 4.74–4.81 (m, 1H), 5.12–5.17 (m, 1H). ¹³C{¹H} NMR (CDCl₃): δ 21.7, 25.0, 26.1, 26.9, 27.2, 27.9, 28.6, 30.8, 31.7, 52.45, 52.51, 57.4, 85.7, 91.4, 171.8 (2C), 206.1. EI-HRMS: calcd for $C_{17}H_{26}O_4$ 294.1831, found 294.1824. Anal. Calcd for $C_{17}H_{26}O_4$: C, 69.36; H, 8.90. Found: C, 69.28; H, 9.05.

Dimethyl 2,2-(2,3-Tridecadiene-1,13-diyl)-1,3-propanedioate (**5g**; n=9). Colorless liquid. Yield: 45.3 mg (42%). ¹H NMR (CDCl₃): δ 0.73–0.90 (m, 1H), 1.14–1.57 (m, 13H), 1.87–2.04 (m, 4H), 2.58 (dd, J=14.5 and 11.6 Hz, 1H), 2.78 (dtd, J=14.5, 4.4, and 0.9 Hz, 1H), 3.72 (s, 3H), 3.73 (s, 3H), 4.63–4.70 (m, 1H), 5.02–5.08 (m, 1H). ¹³C{¹H} NMR (CDCl₃): δ 20.7, 23.2, 23.9, 26.0, 26.37, 26.40, 27.16, 27.18, 31.6, 32.4, 52.5, 52.6, 57.2, 84.5, 91.3, 171.8, 171.9, 206.1. EI-HRMS: calcd for C₁₈H₂₈O₄ 308.1988, found 308.1989. Anal. Calcd for C₁₈H₂₈O₄: C, 70.10; H, 9.15. Found: C, 70.00; H, 9.11.

Dimethyl 2,2-(2,3-Pentadecadiene-1,15-diyl)-1,3-propanedioate (5h; n=11). Colorless liquid. Yield: 38.9 mg (33%). ¹H NMR (CDCl₃): δ 1.07–1.43 (m, 18H), 1.82–2.03 (m, 4H), 2.56 (dd, J = 14.2 and 9.8 Hz, 1H), 2.71 (ddd, J = 14.2, 6.0, and 3.2 Hz, 1H), 3.72 (s, 3H), 3.73(s, 3H), 4.69–4.71 (m, 1H), 5.07–5.13 (m, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 23.6, 24.2, 24.4, 25.9, 26.0, 26.2, 26.3, 27.4, 27.5, 29.0, 31.9, 32.9, 52.41, 52.44, 57.3, 84.6, 91.2, 171.81, 171.84, 206.2. EI-HRMS: calcd for C₂₀H₃₂O₄ 336.2301, found 336.2302.

5,5,13,13-Tetra(methoxycarbonyl)cyclohexadeca-1,2,9,10-tetraene (6a; n=3). Diastereomeric mixture (13 C NMR signals not completely resolved). 17 Colorless liquid. Yield: 23.5 mg (30%). 1 H NMR (CDCl₃): δ 1.02–1.10 (m, 1H), 1.17–1.41 (m, 3H), 1.79–2.12 (m, 8H), 2.49–2.55 (m, 2H), 2.66–2.72 (m, 2H), 3.70–3.72(s, 12H), 4.63–4.81 (m, 2H), 4.97–5.10 (m, 2H). 13 C{ 1 H} NMR (CDCl₃): δ 23.9, 24.8, 29.7, 31.0, 31.1, 31.5, 31.8, 32.1, 52.5, 52.57, 52.64, 57.0, 84.3, 84.8, 90.5, 90.6, 171.51, 171.55, 171.60, 171.62, 205.7, 206.0. EI-HRMS: calcd for C₂₄H₃₂O₈ 448.2097, found 448.2098. Anal. Calcd for C₂₄H₃₂O₈: C, 64.27; H, 7.19. Found: C, 64.08; H, 7.25.

5,5,14,14-Tetra(methoxycarbonyl)cyclooctadeca-1,2,10,11-tetraene (6b; n=4). Diastereomeric mixture (13 C NMR signals not completely resolved). Colorless liquid. Yield: 13.3 mg (16%). HNMR (CDCl₃): δ 0.94–1.70 (m, 8H), 1.82–2.08 (m, 8H), 2.50–2.57 (m, 2H), 2.68–2.73 (m, 2H), 3.72–3.75 (m, 12H), 4.65–4.80 (m, 2H), 5.04–5.15 (m, 2H). NMR (CDCl₃): δ 23.7, 24.1, 28.4, 29.0, 30.2, 30.8, 31.7, 31.9, 32.6, 32.9, 52.50, 52.54, 52.6, 57.0, 57.1, 84.6, 85.5, 90.7, 90.9, 171.6, 171.7, 205.7, 206.0. EI-HRMS: calcd for $C_{26}H_{36}O_8$ 476.2410, found 476.2414. Anal. Calcd for $C_{26}H_{36}O_8$: C, 65.53; H, 7.61. Found: C, 65.81; H, 7.77.

5,5,15,15-Tetra(methoxycarbonyl)cycloeicosa-1,2,11,12-tetraene (6c; n=5). Diastereomeric mixture (13 C NMR signals not completely resolved). Colorless liquid. Yield: 21.2 mg (24%). NMR (CDCl₃): δ 0.96–1.12 (m, 2H), 1.16–1.45 (m, 10H), 1.82–2.07

(m, 8H), 2.51–2.56 (m, 2H), 2.68–2.72 (m, 2H), 3.72–3.73 (m, 12H), 4.65–4.75 (m, 2H), 5.01–5.09 (m, 2H). 13 C{ 1 H} NMR (CDCl $_{3}$): δ 24.0, 24.1, 29.3, 29.5, 29.8, 30.1, 30.6, 30.7, 31.6, 31.7, 32.18, 32.22, 52.55, 52.60, 57.1, 57.2, 84.4, 84.8, 90.8, 90.9, 171.7, 205.8, 205.9. EI-HRMS: calcd for C $_{28}$ H $_{40}$ O $_{8}$ 504.2723, found 504.2724. Anal. Calcd for C $_{28}$ H $_{40}$ O $_{8}$: C, 66.65; H, 7.99. Found: C, 66.49; H, 8.12.

5,5,16,16-Tetra(methoxycarbonyl)cyclodocosa-1,2,12,13-tetraene (6d; n=6). Diastereomeric mixture (13 C NMR signals not completely resolved). 17 Colorless liquid. Yield: 25.2 mg (27%). 14 H NMR (CDCl₃): δ 0.96–1.38 (m, 16H), 1.84–2.02 (m, 8H), 2.50–2.56 (m, 2H), 2.68–2.74 (m, 2H), 3.72–3.73 (m, 12H), 4.65–4.77 (m, 2H), 5.03–5.13 (m, 2H). 13 C{ 14 H NMR (CDCl₃): δ 24.3, 24.4, 29.0, 29.6, 29.87, 29.92, 30.3, 30.4, 30.6, 31.9, 32.6, 32.7, 52.50, 52.54, 57.11, 57.13, 84.6, 85.0, 90.99, 91.04, 171.8, 205.78, 205.82. EI-HRMS: calcd for C₃₀H₄₄O₈: 532.3036, found 532.3037. Anal. Calcd for C₃₀H₄₄O₈: C, 67.64; H, 8.33. Found: C, 67.44; H, 8.38.

5,5,17,17-Tetra(methoxycarbonyl)cyclotetracosa-1,2,13,14-tetraene (**6e**; n=7). Diastereomeric mixture (13 C NMR signals not completely resolved). Colorless liquid. Yield: 18.6 mg (19%). HNMR (CDCl₃): δ 0.97–1.43 (m, 20H), 1.83–2.05 (m, 8H), 2.51–2.57 (m, 2H), 2.66–2.72 (m, 2H), 3.72 (s, 6H), 3.73(s, 6H), 4.66–4.75 (m, 2H), 5.02–5.10 (m, 2H). Clark NMR (CDCl₃): δ 24.1, 24.3, 29.4, 29.7, 29.75, 29.82, 29.9, 30.3, 30.38, 30.43, 30.8, 31.6, 31.7, 31.8, 32.3, 52.5, 52.6, 57.18, 57.23, 84.5, 84.6, 91.0, 91.1, 171.8, 205.8, 205.9. EI-HRMS: calcd for $C_{32}H_{48}O_8$ 560.3349, found 560.3353. Anal. Calcd for $C_{32}H_{48}O_8$: C, 68.54; H, 8.63. Found: C, 68.78; H, 8.83.

5,5,18,18-Tetra(methoxycarbonyl)cyclohexacosa-1,2,14,15-tetraene (6f; n=8). Diastereomeric mixture (13 C NMR signals not completely resolved). Colorless liquid. Yield: 14.4 mg (14%). HNMR (CDCl₃): δ 1.02–1.37 (m, 24H), 1.86–2.03 (m, 8H), 2.50–2.57 (m, 2H), 2.67–2.72 (m, 2H), 3.71–3.73 (m, 12H), 4.68–4.76 (m, 2H), 5.04–5.11 (m, 2H). Colorless liquid. CDCl₃): δ 24.2, 24.3, 29.2, 29.6, 29.8, 30.0, 30.17, 30.21, 30.3, 30.4, 30.5, 30.6, 31.9, 32.6, 52.49, 52.52, 57.2, 84.6, 84.8, 91.1, 171.8, 205.8. EI-HRMS: calcd for C₃₄H₅₂O₈ 588.3662, found 588.3658. Anal. Calcd for C₃₄H₅₂O₈: C, 69.36; H, 8.90. Found: C, 69.30; H, 9.09.

5,5,19,19-Tetra(methoxycarbonyl)cyclooctacosa-1,2,15,16-tetraene (**6g**; n=9). Diastereomeric mixture (13 C NMR signals not completely resolved). ¹⁷ Colorless liquid. Yield: 22.7 mg (21%). 14 H NMR (CDCl₃): δ 1.01–1.37 (m, 28H), 1.86–2.05 (m, 8H), 2.56 (dd, J=14.3 and 9.8 Hz, 2H), 2.68 (ddd, J=14.3, 6.1, and 3.3 Hz, 2H), 3.71 (s, 6H), 3.72(s, 6H), 4.68–4.76 (m, 2H), 5.04–5.10 (m, 2H). 13 C{ 14 H} NMR (CDCl₃): δ 24.0, 24.1, 29.3, 29.5, 29.67, 29.72, 29.8, 29.97, 30.03, 30.1, 30.2, 30.3, 30.5, 31.8, 32.4, 52.5, 57.2, 84.58, 84.64, 91.1, 171.8, 205.8. EI-HRMS: calcd for C₃₆H₅₆O₈ 616.3975, found 616.3978. Anal. Calcd for C₃₆H₅₆O₈: C, 70.10; H, 9.15. Found: C, 70.08: H, 9.01.

5,5,21,21-Tetra(*methoxycarbonyl*)*cyclodotriaconta-1,2,17,18-tetraene* (*6h*; *n* = 11). Diastereomeric mixture (13 C NMR signals not completely resolved). Colorless liquid. Yield: 11.8 mg (10%). HNMR (CDCl₃): δ 1.04–1.41 (m, 36H), 1.86–2.05 (m, 8H), 2.55 (ddd, J = 14.3, 9.6, and 1.2 Hz, 2H), 2.67 (ddd, J = 14.3, 6.2, and 3.3 Hz, 2H), 3.71 (s, 6H), 3.72(s, 6H), 4.70–4.78 (m, 2H), 5.04–5.10 (m, 1H). C{}^{1}H NMR (CDCl₃): δ 24.0, 29.20, 29.24, 29.57, 29.63, 29.7, 29.8, 29.85, 29.94, 30.02, 30.07, 30.2, 30.3, 31.8, 32.4, 52.5, 57.3, 84.7, 91.1, 171.9, 205.8. EI-HRMS: calcd for C₄₀H₆₄O₈ 672.4601, found 672.4602. Anal. Calcd for C₄₀H₆₄O₈: C, 71.39; H, 9.59. Found: C, 70.96; H, 9.56.

5,5,13,13,21,21-Hexa(methoxycarbonyl)cyclotetracosa-1,2,9,10,17,18-hexaene (7a; n=3). Diastereomeric mixture (13 C NMR signals not completely resolved). Colorless liquid. Yield: 7.8 mg (10%). H NMR (CDCl₃): δ 1.19–1.35 (m, 6H), 1.86–2.08 (m, 12H), 2.56–2.67 (m, 6H), 3.71–3.74 (m, 18H), 4.79–4.871 (m, 3H), 4.98–5.06 (m, 3H). C(11 H NMR (CDCl₃): δ 23.9, 29.4, 31.4, 32.6, 52.5, 57.3, 84.8, 90.3, 171.6, 205.8, 205.9, 206.0. EI-HRMS: calcd for C₃₆H₄₈O₁₂ 672.3146, found 672.3148.

5,5,15,15,25,25-Hexa(methoxycarbonyl)cyclotriaconta-1,2,11,12,21,22-hexaene (7c; n=5). Diastereomeric mixture (13 C NMR signals not completely resolved). 17 Colorless liquid. Yield: 6.2

mg (7%). ¹H NMR (CDCl₃): δ 1.11–1.43 (m, 18H), 1.86–1.98 (m, 12H), 2.55–2.66 (m, 6H), 3.71–3.72 (m, 18H), 4.78–4.85 (m, 3H), 5.03–5.08 (m, 3H). ¹³C{¹H} NMR (CDCl₃): δ 23.6, 28.7, 28.9, 29.2, 31.8, 32.5, 52.5, 57.5, 84.9, 90.8, 171.8, 205.8. EI-HRMS: calcd for C₄,H₆₀O₁, 756.4085, found 756.4085.

5,5,16,16,27,27-Hexa(methoxycarbonyl)cyclotritriaconta-1,2,12,13,23,24-hexaene (7d; n=6). Diastereomeric mixture (13 C NMR signals not completely resolved). 17 Colorless liquid. Yield: 8.4 mg (9%). 14 H NMR (CDCl₃): δ 1.10–1.41 (m, 24H), 1.86–1.98 (m, 12H), 2.55–2.66 (m 6H), 3.715 (s, 9H), 3.718 (s, 9H), 4.79–4.86 (m, 3H), 5.03–5.09 (m, 3H). 13 C{ 1 H} NMR (CDCl₃): δ 23.8, 28.7, 28.8, 29.1, 29.6, 31.9, 32.5, 52.5, 57.5, 84.9, 90.9, 171.8, 205.8. EI-HRMS: calcd for C₄₅H₆₆O₁₂ 798.4554, found 798.4552.

2,2-(Dodecadiene-1,12-diyl)-1,3-benzodithiole 1,1,3,3-Tetraoxide (10f). To a mixture of $Pd_2(dba)_4$ (5.5 mg, 11 μ mol), dpbp (6.0 mg, 12 μ ol), and KO'Bu (23.6 mg, 210 μ mol) in THF (150 mL) was added 4f (96.9 mg, 210 μ mol, 1.4 mM) by means of syringe under nitrogen. After the mixture was stirred for 120 h at 30 °C, the reaction mixture was concentrated, filtered through a short pad of silica gel, and evaporated to dryness. The crude product was purified by recycle HPLC to afford 10f (53.2 mg, 140 µmol, 67%) as a colorless liquid. ¹H NMR (CDCl₃): δ 1.21–1.86 (m, 12H), 2.02–2.15 (m, 2H), 2.26– 2.38 (m, 2H), 2.88 (dt, J = 15.3 and 4.3 Hz, 1H), 2.99 (dd, J = 15.3and 10.6 Hz, 1H), 5.22-5.28 (m, 1H), 5.43-5.50 (m, 1H) 7.88-7.93 (m, 2H), 7.98-8.04 (m, 2H). ${}^{13}C{}^{1}H$ NMR (CDCl₂): δ 21.1, 25.3, 25.7, 26.1, 26.6, 26.7, 27.5, 29.0, 29.7, 79.5, 83.4, 92.6, 123.1, 123.2, 135.1, 135.2, 135.8, 136.3, 207.1. EI-HRMS: calcd for C₁₉H₂₄O₄S₂ 380.1116, found 380.1121. Anal. Calcd for C₁₉H₂₄O₄S₂: C, 59.97; H, 6.36. Found: C, 60.04; H, 6.36.

Palladium-Catalyzed Asymmetric Synthesis of Axially Chiral 5f and 10f. The asymmetric reactions were conducted in the same way as with the nonasymmetric reaction described above except a chiral ligand was used instead of achiral dpbp (see Table 3 for details). (*S*)-**5f** (entry 7, Table 3): $[\alpha]^{25}_D = +93.1$ (c 0.73, CHCl₃ for the sample of (*S*)-70% ee). Chiral HPLC analysis conditions: Chiralpak IC; eluent, hexane/'PrOH = 200/1; flow rate, 1.0 mL/min; t_1 [(R)-enantiomer] = 20.8 min, t_2 [(R)-enantiomer] = 40.7 min. (R)-10f (entry 10, Table 3): $[\alpha]^{25}_D = +56.9$ (R) (R)-6.37, CHCl₃ for the sample of (R)-51% ee). Chiral HPLC analysis conditions: Chiralpak IC; eluent, hexane/'PrOH = 9/1; flow rate, 1.0 mL/min; ; R1 [(R)-enantiomer] = 20.0 min, R3 [(R)-enantiomer] = 31.9 min.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.7b01204.

 $^{1}\text{H-}$ and ^{13}C NMR spectra for all the new compounds and chiral HPLC chromatograms of (+)-5f and (+)-10f (PDF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) (a) Taylor, D. R. Chem. Rev. 1967, 67, 317. (b) Pasto, D. J. Tetrahedron 1984, 40, 2805. (c) Patai, S., Ed. The Chemistry of Ketenes, Allenes, and Related Compounds; Wiley: Chichester, 1980. (d) Landor, S. R., Ed. The Chemistry of the Allenes; Academic Press: London, 1982. (e) Coppola, G. M.; Schuster, H. F. Allenes in Organic Synthesis; Wiley: New York, 1984. (f) Krause, N., Hashmi, A. S. K., Eds. Modern Allene Chemistry; Wiley-VCH: Weinheim, 2004.
- (2) (a) Krause, N., Ed. Science of Synthesis: Houben-Weyl Methods of Molecular Transformations; Georg Thieme Verlag: Stuttgart, 2008; Vol. 44 (Cumulenes and Allenes). (b) Brummond, K. M.; DeForrest, J. E. Synthesis 2007, 2007, 795. (c) Ogasawara, M. Tetrahedron: Asymmetry 2009, 20, 259. (d) Yu, S.; Ma, S. Chem. Commun. 2011, 47, 5384. (e) Yu, S.; Ma, S. Angew. Chem., Int. Ed. 2012, 51, 3074. (f) Ye, J.; Ma, S. Org. Chem. Front. 2014, 1, 1210.
- (3) (a) Johnson, R. P. Chem. Rev. 1989, 89, 1111. (b) Hopf, H. In The Chemistry of Ketenes, Allenes, and Related Compounds; Patai, S., Ed.; Wiley: Chichester, 1980; p 823. (c) Coppola, G. M.; Schuster, H. F. Allenes in Organic Synthesis; Wiley: New York, 1984; p 38. (d) Christl, M. In Modern Allene Chemistry; Krause, N., Hashmi, A. S. K., Eds.; Wiley-VCH: Weinheim, 2004; p 243. (e) Kawase, T. In Science of Synthesis: Houben-Weyl Methods of Molecular Transformations; Krause, N., Ed.; Georg Thieme Verlag: Stuttgart, 2008; Vol. 44 (Cumulenes and Allenes), Chapter 44.3, p 395.
- (4) (a) Skattebøl, L. Tetrahedron Lett. 1961, 2, 167. (b) Moore, W. R.; Ward, H. R. J. Org. Chem. 1962, 27, 4179. (c) Untch, K. G.; Martin, D. J.; Castellucci, N. T. J. Org. Chem. 1965, 30, 3572.
- (5) (a) Ball, W. J.; Landor, S. R. J. Chem. Soc. 1962, 2298. (b) Moore, W. R.; Bertelson, R. C. J. Org. Chem. 1962, 27, 4182.
- (6) Zelder, C.; Krause, N. Eur. J. Org. Chem. 2004, 2004, 3968.
- (7) Janßen, C. E.; Krause, N. Eur. J. Org. Chem. **2005**, 2005, 2322.
- (8) (a) Brody, M. S.; Williams, R. M.; Finn, M. G. J. Am. Chem. Soc. 1997, 119, 3429. (b) Kulyk, S.; Dougherty, W. G., Jr.; Kassel, W. S.; Fleming, S. A.; Sieburth, S. M. Org. Lett. 2010, 12, 3296. (c) Mömming, C. M.; Kehr, G.; Wibbeling, B.; Fröhlich, R.; Schirmer, B.; Grimme, S.; Erker, G. Angew. Chem., Int. Ed. 2010, 49, 2414.
- (9) (a) Ogasawara, M.; Ikeda, H.; Hayashi, T. Angew. Chem., Int. Ed. 2000, 39, 1042. (b) Ogasawara, M.; Okada, A.; Nakajima, K.; Takahashi, T. Org. Lett. 2009, 11, 177. (c) Ogasawara, M.; Murakami, H.; Furukawa, T.; Takahashi, T.; Shibata, N. Chem. Commun. 2009, 7366. (d) Ogasawara, M.; Suzuki, M.; Takahashi, T. J. Org. Chem. 2012, 77, 5406 and references cited therein.
- (10) (a) Ogasawara, M.; Ikeda, H.; Nagano, T.; Hayashi, T. *J. Am. Chem. Soc.* **2001**, 123, 2089. (b) Ogasawara, M.; Okada, A.; Subbarayan, V.; Sörgel, S.; Takahashi, T. *Org. Lett.* **2010**, 12, 5736 and references cited therein..
- (11) Ogasawara, M.; Okada, A.; Watanabe, S.; Fan, L.; Uetake, K.; Nakajima, K.; Takahashi, T. *Organometallics* **2007**, *26*, 5025.
- (12) Daoust, K. J.; Hernandez, S. M.; Konrad, K. M.; Mackie, I. D.; Winstanley, J., Jr.; Johnson, R. P. J. Org. Chem. **2006**, 71, 5708.
- (13) Price, J. D.; Johnson, R. P. Tetrahedron Lett. 1986, 27, 4679.
- (14) (a) Desai, N. B.; McKelvie, N.; Ramirez, F. J. Am. Chem. Soc. **1962**, 84, 1745. (b) Corey, E. J.; Fuchs, P. L. Tetrahedron Lett. **1972**, 13, 3769.
- (15) For palladium-catalyzed selective substitutions of one of the two halides in 1,1-dihalo-1-alkenes, see: (a) Minato, A.; Suzuki, K.; Tamao, K. J. Am. Chem. Soc. 1987, 109, 1257. (b) Roush, W. R.; Koyama, K.; Curtin, M. L.; Moriarty, K. J. J. Am. Chem. Soc. 1996, 118, 7502. (c) Uenishi, J.; Kawahama, R.; Yonemitsu, O.; Tsuji, J. J. Org. Chem. 1996, 61, 5716. (d) Shen, W.; Wang, L. J. Org. Chem. 1999, 64, 8873. (e) Zeng, X.; Hu, Q.; Qian, M.; Negishi, E. J. Am. Chem. Soc. 2003, 125, 13636.
- (16) dpbp = 2,2'-bis(diphenylphosphino)-1,1'-biphenyl. See: Ogasawara, M.; Yoshida, K.; Hayashi, T. *Organometallics* **2000**, *19*, 1567 and references cited therein.
- (17) The *dl/meso-*diastereomeric ratios could not be determined due to the incomplete separation of their NMR signals.
- (18) DPEphos = bis[2-(diphenylphosphino)phenyl]ether. See: Kranenburg, M.; van der Burgt, Y. E. M.; Kamer, P. C. J.; van

- Leeuwen, P. W. N. M.; Goubitz, K.; Fraanje, J. Organometallics 1995, 14, 3081.
- (19) Jacobs, T. L.; McClenon, J. R.; Muscio, O. J., Jr. J. Am. Chem. Soc. 1969, 91, 6038.
- (20) (a) Lowe, G. Chem. Commun. 1965, 411. (b) Brewster, J. H. Topics in Stereochemistry 1967, 2, 1.
- (21) Coulson, D. R. Inorg. Synth. 1972, 13, 121.
- (22) Ukai, T.; Kawazura, H.; Ishii, Y.; Bonnet, J. J.; Ibers, J. A. J. Organomet. Chem. 1974, 65, 253.
- (23) Saito, T.; Yokozawa, T.; Ishizaki, T.; Moroi, T.; Sayo, N.; Miura, T.; Kumobayashi, H. Adv. Synth. Catal. 2001, 343, 264.
- (24) Hu, A.; Ogasawara, M.; Sakamoto, T.; Okada, A.; Nakajima, K.; Takahashi, T.; Lin, W. Adv. Synth. Catal. 2006, 348, 2051.
- (25) Cai, D.; Payack, J. F.; Bender, D. R.; Hughes, D. L.; Verhoeven, T. R.; Reider, P. J. J. Org. Chem. 1994, 59, 7180.
- (26) Hu, A.; Ngo, H. L.; Lin, W. Angew. Chem., Int. Ed. 2004, 43, 2501.
- (27) Kündig, E. P.; Cunningham, A. F. Tetrahedron 1988, 44, 6855.
- (28) Clyne, D. S.; Weiler, L. Tetrahedron 1999, 55, 13659.
- (29) Somekawa, K.; Okuhira, H.; Sendayama, M.; Suishu, T.; Shimo, T. J. Org. Chem. 1992, 57, 5708.
- (30) LeBorgne, J.-F. J. Organomet. Chem. 1976, 122, 123.
- (31) Zakrzewski, J.; Grodner, J.; Bobbitt, J. M.; Karpinska, M. Synthesis 2007, 2007, 2491.
- (32) Bestmann, H. J.; Kellermann, W. Synthesis 1994, 1994, 1257.
- (33) Porter, N. A.; Chang, V. H.-T.; Magnin, D. R.; Wright, B. T. J. Am. Chem. Soc. 1988, 110, 3554.